

## **Appendix D: EPA's Legal Authority for Gasoline Sulfur Control**

We are adopting gasoline sulfur controls pursuant to our authority under Section 211(c)(1) of the Clean Air Act. This section gives us the authority to “control or prohibit the manufacture, introduction into commerce, offering for sale, or sale” of any fuel or fuel additive (A) whose emission products, in the judgment of the Administrator, cause or contribute to air pollution “which may be reasonably anticipated to endanger the public health or welfare” or (B) whose emission products “will impair to a significant degree the performance of any emission control device or system which is in general use, or which the Administrator finds has been developed to a point where in a reasonable time it would be in general use” were the fuel control or prohibition adopted. The following sections describe our current our requirements that affect gasoline sulfur content, and explain our bases for controlling gasoline sulfur under Section 211(c)(1).

### **A. EPA's Current Regulatory Requirements for Gasoline**

We currently have regulatory requirements for conventional and reformulated gasoline (RFG), adopted under Sections 211(c) and 211(k) of the Act. RFG is required to be sold in certain ozone nonattainment areas. Gasoline sold in the rest of the country is subject to the conventional gasoline requirements. See 40 CFR part 80.

Both the RFG and conventional gasoline (CG) requirements include a NO<sub>x</sub> performance standard that requires refiners to achieve a certain level of NO<sub>x</sub> control compared to 1990 baseline levels. As a practical matter, compliance with this performance standard results in limiting sulfur levels in RFG. The NO<sub>x</sub> reductions required by the Phase 2 RFG requirements, effective on January 1, 2000, are expected to result in RFG sulfur levels of about 150 ppm. In addition, EPA's regulations require compliance with the RFG and CG standards (including the NO<sub>x</sub> performance standard) to be calculated using the Complex Model beginning in 1998. This model contains range limits for RFG for a number of fuel parameters that affect NO<sub>x</sub> performance, including a range of zero to 500 ppm for sulfur. Therefore, the requirement to use the Complex Model effectively limits sulfur levels in RFG to no more than 500 ppm. The sulfur Complex Model range limit for RFG is the only direct regulation of sulfur content under Section 211(c)(1). However, the NO<sub>x</sub> performance standards for RFG and CG have an indirect effect on

sulfur content.<sup>1</sup>

All gasoline is subject to Section 211(f) of the Act, which prohibits fuel or fuel additive manufacturers from introducing into commerce, or increasing the concentration in use of, any fuel or fuel additive for general use in light duty motor vehicles which is not “substantially similar” to the fuel used in the certification of model year 1975 or later vehicles or engines. We have interpreted “substantially similar” for unleaded gasoline to include any gasoline meeting the 1988 ASTM specifications for unleaded gasoline (ASTM D 4814-88<sup>2</sup>), which limits the sulfur content of unleaded gasoline to 0.1 weight percent (1000 ppm) sulfur.

### **B. How the Gasoline Sulfur Control Program Meets the CAA Section 211(c) Criteria**

Under Section 211(c)(1), EPA may adopt a fuel control if at least one of the following two criteria is met: 1) the emission products of the fuel cause or contribute to air pollution which may reasonably be anticipated to endanger public health or welfare, or 2) the emission products of the fuel will significantly impair emissions control systems in general use or which would be in general use were the fuel control to be adopted. We are promulgating controls on sulfur levels in gasoline based on both of these criteria. Under the first criterion, we believe that emissions products of sulfur in gasoline used in Tier 1 and LEV technology vehicles contribute to ozone pollution, air toxics, and PM. Under the second criterion, we believe that gasoline sulfur in fuel that will be used in Tier 2 technology vehicles will significantly impair the emissions control systems expected to be used in such vehicles. The following sections summarize our analysis of each criterion.

#### **1. Health and Welfare Concerns of Air Pollution Caused by Sulfur in Gasoline**

We believe that the emission products of gasoline sulfur contribute to air pollution that can reasonably be anticipated to endanger public health and welfare. The combustion products of the sulfur-containing compounds in gasoline (SO<sub>2</sub> and other sulfur oxides) contribute to air pollution that has adverse impacts on public health and welfare. The greatest impact of gasoline sulfur on pollution is the increase in emissions of hydrocarbons (including hazardous air pollutants such as benzene and 1,3-butadiene), NO<sub>x</sub>, particulate matter, and compounds such as nitrates and sulfates that become particulates in the atmosphere. As explained below, and in

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<sup>1</sup>Because sulfur is directly or indirectly controlled by EPA requirements, and will be controlled directly under today’s action, states are preempted from initiating sulfur control programs unless they are identical to the federal requirements. See the discussion in Section IV.C. of the preamble on this subject.

<sup>2</sup>Standard Specification for Automotive Spark-Ignition Engine Fuel

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more detail in Chapter III above, these increased emissions result primarily from the adverse impact of high sulfur levels on the automotive catalysts used in the vehicles which have recently entered the fleet or will be used to comply with the Tier 2 standards. The health and welfare implications of the emissions of these compounds are discussed in greater detail in Section III of the Preamble.

Section 211(c)(2)(A) requires that, prior to adopting a fuel control based on a finding that the fuel's emission products contribute to air pollution that can reasonably be anticipated to endanger public health or welfare, EPA consider "all relevant medical and scientific evidence available, including consideration of other technologically or economically feasible means of achieving emission standards under [section 202 of the Act]."

EPA's analysis of the medical and scientific evidence relating to the emissions impact of sulfur in gasoline is described in more detail in the RIA.

EPA has also satisfied the statutory requirement to consider "other technologically or economically feasible means of achieving emission standards under section [202 of the Act]." This provision has been interpreted as requiring consideration of establishing emissions standards under § 202 prior to establishing controls or prohibitions on fuels or fuel additives under § 211(c)(1)(A). See *Ethyl Corp. v. EPA*, 541 F.2d. 1, 31-32 (D.C. Cir. 1976). In *Ethyl*, the court stated that § 211(c)(2)(A) calls for good faith consideration of the evidence and options, not for mandatory deference to regulation under § 202 compared to fuel controls. *Id.* at 32, n.66.

In today's action, EPA for the first time is establishing standards for fuels and vehicles together. Thus, it is first important to consider that the sulfur standards are not being adopted as an alternative to vehicle emissions standards, but in addition to such standards, and as a necessary prerequisite to ensuring that vehicles can meet the vehicle standards. In addition, the Tier 2 standards being adopted today will begin phasing-in in 2004, but will not be fully phased in for the fleet of new motor vehicles until 2009, and even at that time, many non-Tier 2 vehicles will still be on the road. Thus, another point to consider is that the emissions standards under § 202 will achieve smaller emissions benefits in the early years of the program and will not achieve their full emissions benefits for a number of years, while the sulfur standards will achieve significant, immediate emissions benefits through reducing emissions from the existing fleet of motor vehicles (primarily Tier 1 vehicles and LEVs), and will continue to achieve increasing benefits as the fleet turns over to Tier 2 vehicles, especially in light of the expected increase in vehicle miles travelled.

EPA has also considered emissions standards under § 202 that are more stringent than Tier 2 as an alternative to regulating gasoline sulfur. However, for the reasons described in Chapter IV, we conclude that the Tier 2 standards represent the level of emission control that is economically and technologically feasible from new motor vehicles in the time frame over which the standards are implemented. Moreover, we considered Tier 2 standards without control of

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gasoline sulfur as an alternative to regulating gasoline sulfur. However, we have concluded that the Tier 2 standards would not be feasible without control of gasoline sulfur. For these reasons, we find that the alternatives of either more stringent vehicle standards, or Tier 2 standards without sulfur control, are not technologically or economically feasible options to regulating gasoline sulfur.

Our consideration of other technologically and economically feasible means of achieving emissions standards under § 202 of the Act supports the conclusion that the sulfur standards adopted today represent an appropriate exercise of the Agency's discretion under § 211(c)(1)(A), even when the Tier 2 standards are considered.

### **2. Impact of Gasoline Sulfur Emission Products on Emission Control Systems**

EPA believes that sulfur in gasoline can significantly impair the emissions control technology of vehicles designed to meet the Tier 2 emissions standards. We know that gasoline sulfur has a negative impact on vehicle emission controls. This is not a new development. Vehicles depend on the catalytic converter to oxidize or reduce emissions of HC, CO, and NO<sub>x</sub>. Sulfur and sulfur compounds attach or "adsorb" to the precious metals which are required to convert these emissions. Sulfur also blocks sites on the catalyst designed to store oxygen which are necessary to optimize NO<sub>x</sub> emissions conversions. While the amount of sulfur contamination can vary depending on the metals used in the catalyst and other aspects of the design and operation of the vehicle, some level of sulfur contamination will occur in any catalyst.

For older vehicles designed to meet Tier 0 and Tier 1 emission standards, this sulfur contamination increases emissions of NMHC and NO<sub>x</sub> by almost 17 percent when one of these vehicles is operated on gasoline containing 330 ppm sulfur (approximately the current national average sulfur level) compared to operation on gasoline with 30 ppm sulfur (which is close to California's current average sulfur level, and is the average sulfur level adopted today). Thus, Tier 0 and Tier 1 vehicles have higher emissions when they are exposed to sulfur levels substantially higher than the sulfur standard adopted today. This increase is generally not enough to cause a vehicle to exceed the full useful life emission standards in practice, because car manufacturers design the vehicles with a margin of safety to compensate for deterioration in emissions performance over the life of the vehicle. However, it does lead to greater in-use emissions than achieved with the gasoline sulfur control.

The sulfur impact on the catalysts used in later model vehicles is clearly significant. High sulfur levels have been shown to significantly reduce the conversion efficiency of the emissions control systems of cleaner, later technology vehicles. The California LEV standards and Federal NLEV standards, as well as California's new LEV-II standards and our Tier 2 standards, require catalysts to be extremely efficient to adequately reduce emissions over the full useful life of the vehicle. Recent test programs conducted by the automotive and oil industries show that LEV and

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ULEV vehicles can experience, on average, a 100 percent increase in NMHC and 197 percent increase in NO<sub>x</sub> emissions when operated on 330 ppm sulfur fuel compared to 30 ppm sulfur fuel (see Appendix B for more details). This level of emissions increase is significant enough that it would undermine the technical and economic feasibility of the Tier 2 standards.

This level of impact on emission control system efficiency would mean actual in-use emissions reductions from the Tier 2 standards would be undercut to such a degree that the resulting limited in-use emissions reductions would not warrant the expense imposed by the Tier 2 standards, and would not achieve the in-use emissions reductions from these motor vehicles needed to address the air quality problems described elsewhere in this notice. In addition, the concerns about irreversibility of the damage to the catalyst mean it would not be feasible to design an emission control system that would offset this level of impact on the efficiency of the control system in order to comply over the useful life of the vehicles. Average sulfur levels in the U.S. are currently high enough to significantly impair the emissions control systems in new technology vehicles, and to potentially cause these vehicles to fail emission standards required for vehicles up through 100,000 miles (or more) of operation.

Sulfur in gasoline can also significantly impair the onboard diagnostic (OBD) systems in current and future vehicles. EPA regulations require all vehicles to be equipped with OBD systems that monitor catalyst performance and other emissions-related performance, and warns the vehicle owner if the emissions control system is not functioning properly. In a 1997 staff paper, EPA concluded that sulfur in gasoline can directly impact OBD systems by affecting the OBD system's oxygen sensors.<sup>1</sup> It is possible that high sulfur levels may impair the OBD system in such a way that it does not recognize an improperly functioning catalyst, and fails to warn the owner. In addition, it is not clear that the conditions which may reverse some of sulfur's effect on the catalyst will also reverse this impact on the OBD system's oxygen sensors. The impact of sulfur on OBD systems in cleaner technology vehicles may be even more significant, since the OBD malfunction thresholds are expressed as multiples of the applicable hydrocarbon standard. Therefore, the impact of sulfur on OBD systems in vehicles meeting more stringent hydrocarbon standards would be more significant in relative terms.

### **3. Sulfur Levels that Tier 2 Vehicles Can Tolerate**

We believe that Tier 2 vehicles that operate on gasoline will, on average over their long-term operation, have to use fuel with sulfur levels no greater than 30 ppm to avoid significant impairment of their emissions control systems. Furthermore, short-term operation on gasoline with sulfur levels higher than 80 ppm will have a significant adverse effect on the desired emission performance and will significantly impair the emissions control system. These conclusions are based on data collected on vehicles currently sold in California or being developed for sale in California and the Northeast (the latter under the NLEV program).

The test data from industry test programs and individual automotive and catalyst

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manufacturers show that even very low levels of sulfur have some impact on catalyst performance. The data also show that the greatest increase in emissions comes as the sulfur level is increased from the lowest levels. At higher sulfur levels, the catalyst is approaching the point of being saturated with sulfur, and its performance is already impaired, such that an additional increase in sulfur content has a smaller impact on emissions. This trend applies generally for all of the regulated pollutants (NMHC, CO, NO<sub>x</sub>). However, for most vehicles, the impact is greatest for NO<sub>x</sub>.

While the overall trends demonstrate that high sulfur levels significantly impair the emissions control system of newer technology vehicles, the data also shows that some vehicles are much less sensitive to sulfur than others. The reasons for these vehicle-to-vehicle variations are not fully understood. We have identified a number of factors involved in the vehicle design and operation which appear to contribute to the variation. To summarize briefly, sulfur sensitivity is impacted not only by the catalyst formulation (the types and amounts of precious metals used in the catalyst) but also by the following factors:

- the materials used to provide oxygen storage capacity in the catalyst, as well as the general design of the catalyst,
- the location of the catalyst relative to the engine, which impacts the temperatures inside the catalyst,
- the mix of air and fuel entering the engine over the course of operation, which is varied by the engine's computer in response to the driving situation and affects the mix of gases entering the catalyst from the engine, and
- the speeds the car is driven at and the load the vehicle is carrying, which also impact the temperatures experienced by the catalyst.

All of these factors contribute not only to the degree to which sulfur will poison a catalyst, but also whether and how easily the sulfur will be removed during a vehicle's normal operation. This cycle of sulfur collection (adsorption) and removal (desorption) in the catalyst is what ultimately affects sulfur's net impact on emissions and the emissions control system, both short and long term. Since these factors vary for every vehicle, the sulfur impact varies for every vehicle to some degree. There is no single factor that guarantees that a vehicle will be very sensitive or very insensitive to sulfur. None of the data that we have reviewed indicates a vehicle design which is completely insensitive to sulfur, or even capable of tolerating average sulfur levels above 30 ppm without a significant impairment of its emissions control system.

Therefore, based on the data and information obtained from catalyst manufacturers, we have also concluded that there are no viable emission control alternatives that could achieve the

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same level of emission standards without reducing commercial gasoline sulfur levels, as explained in the next section.

In summary, we have concluded that the sensitivity of automotive catalysts to sulfur has increased to such a degree in vehicle technology currently available, and expected to be used to meet the Tier 2 standards, that sulfur levels in gasoline must be reduced to enable these catalysts to operate properly. Not only will harmful emissions from vehicles on the road today be reduced through lowering gasoline sulfur levels, but the emissions control systems expected to be used to attain the Tier 2 standards will be significantly impaired if sulfur levels are not substantially reduced from current levels. A lesser reduction in gasoline sulfur levels nationwide would likely require us to reduce the stringency of the Tier 2 standards. While the impact on emissions control systems of Tier 2 vehicles and LEVs is a sufficient basis to control gasoline sulfur under Section 211(c)(1)(B), a similar analysis for Tier 0 and Tier 1 vehicles also supports a determination that gasoline sulfur levels significantly impair the emissions control systems of these vehicles. This is because the effect of sulfur in reducing catalyst efficiency and thereby increasing emissions exists for all vehicles at issue here (Tier 0 through Tier 2), presenting more a question of difference in degree than in the nature of the effects.

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### **4. Sulfur Sensitivity of Other Catalysts**

Section 211(c)(2)(B) requires that, prior to adopting a fuel control based on a significant impairment to vehicle emissions control systems, EPA consider available scientific and economic data, including a cost benefit analysis comparing emissions control devices or systems which are or will be in general use that require the fuel control with such devices or systems which are or will be in general use that do not require the fuel control. As described above, we conclude that the emissions control systems expected to be used to meet the Tier 2 standards would be significantly impaired by operation on high sulfur gasoline. Our analysis of the available scientific and economic data can be found in the Preamble, and Chapters IV, V, and VII above, including an analysis of the environmental benefits of the fuel control, an analysis of the costs and the technological feasibility of controlling sulfur to the levels required by today's action, and a cost-benefit analysis of the sulfur control and Tier 2 vehicle emissions standards. Under Section 211(c)(2)(B), EPA is also required to compare the costs and benefits of emissions control systems that are not sulfur-sensitive, if any such systems are or are will be in general use.

We have determined that there are not (and will not be in the foreseeable future) emission control devices available for general use in gasoline-powered vehicles that can meet the Tier 2 emission standards and would not be significantly impaired by gasoline with high sulfur levels. All catalysts are sensitive to sulfur to some degree. As explained in Section IV.A of the Preamble, as well as in Appendix B above, we cannot identify one or more factors that definitively determine sulfur sensitivity, because sulfur sensitivity seems to be due to a

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combination of many factors that vary by vehicle. Hence, it is not possible to identify alternative designs that can tolerate existing (or even intermediate) sulfur levels and that can reasonably be expected to be applied to all cars and light trucks meeting Tier 2 standards.

As described in Section IV.A. of the Preamble, we anticipate that all the gasoline vehicle technologies expected to be used to meet the Tier 2 standards will require the use of low sulfur gasoline. If we do not control gasoline sulfur to the levels required by today's rule, we would not be able to set Tier 2 standards as stringent as those we are adopting. Moreover, vehicles already on the road would continue to emit at higher levels than they would if operated on low sulfur fuel. These benefits from the existing fleet, which are particularly significant in the early years of the program, cannot be achieved through new vehicle emissions standards. Consequently, we conclude that the benefits that would be achieved through implementation of the vehicle and sulfur control programs in today's rule cannot be achieved through the use of emission control technology that is not sulfur-sensitive.

This also means that if we were to adopt vehicle emissions control standards without controlling gasoline sulfur content, the standards would be significantly less stringent than those based on what would be technologically feasible with current sulfur levels, rather than what is feasible with lower sulfur. In such a situation, the cost of the vehicle emissions control technology would likely be similar to the costs of meeting the Tier 2 standards, because the same technologies would be used. However, the emissions benefits of those technologies would be significantly less than what would be achieved with low sulfur gasoline, because the emissions control technology for gasoline vehicles currently in use, and expected to be used in the future, would be significantly impaired by high sulfur fuel. Thus, the same benefits achieved by today's program could not be achieved through vehicle emissions standards alone, because of the sensitivity of the emissions control technology to sulfur.

### **5. Effect of Gasoline Sulfur Control on the Use of Other Fuels or Fuel Additives**

Section 211(c)(2)(C) requires that, prior to prohibiting a fuel or fuel additive, EPA establish that such prohibition will not cause the use of another fuel or fuel additive "which will produce emissions which endanger the public health or welfare to the same or greater degree" than the prohibited fuel or additive. This finding is required by the Act only prior to prohibiting a fuel or additive, not prior to controlling a fuel or additive. Since EPA is not prohibiting sulfur in gasoline, but rather controlling the levels of sulfur in gasoline, this finding is not required prior to regulation. However, EPA does not believe that the sulfur standards adopted today will result in the use of any other fuel or additive that will produce emissions that will endanger public health or welfare to the same or greater degree as the emissions produced by gasoline with current sulfur levels.

We believe that gasoline formulated to meet the low sulfur standards will have a

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significant net benefit to public health due to reduced emissions of harmful compounds. Other changes to the composition of the gasoline are likely to accompany the reduction in sulfur content. While some of these changes may involve increases in the content of certain compounds that tend to lead to more harmful emission products from the engine itself, we believe that the improved catalyst performance enabled by the low sulfur fuel will more than offset any slight increase in harmful emissions from the engine that might result, such that the net emissions effect of the sulfur control is a substantial reduction in emissions, compared to emissions from gasoline without the sulfur control.

It is difficult to quantify this impact because it depends on the specific approaches that each refiner takes to reduce their gasolines' sulfur levels, as well as the composition of the gasoline overall. However, some general trends can be identified, and based on these trends we have drawn the conclusion that low sulfur gasoline will pose no new, increased risk to human health relative to the higher sulfur gasoline it replaces.

Some refiners already make gasolines that meet the sulfur standards. Others will make modest changes in the way in which they blend refinery streams to produce low sulfur gasoline. But most refiners will have to install some desulfurization technology and/or otherwise substantially change their operation. If a refiner chooses a traditional route to desulfurize gasoline, he will likely select a desulfurization technology which has the undesirable side effect of reducing the octane content of the gasoline streams. To make up that octane, the refiner has several options. All of these options, whether increasing the aromatics or olefins content of the gasoline through other processing changes, or through the addition of oxygenates such as ethanol or MTBE, could lead to increased emissions of air toxics (benzene, 1,3-butadiene, aldehydes) if the emissions performance of the vehicle catalyst remained constant. However, since low sulfur gasoline will enable very low emitting catalysts and will improve the performance of existing catalysts, the catalyst will be able to convert these toxic emissions into less harmful compounds. Because of the diversity among refineries, it is impossible to estimate with any certainty how many refiners may choose this route.

If a refiner chooses one of the improved technologies for sulfur removal, the technologies on which much of our economic analysis for this action is based (as discussed in Sections IV.C and IV.D of the Preamble), there will be less of a need to increase high octane compounds in the gasoline. These improved technologies are designed to reduce the octane loss that occurs with the traditional technologies. Because the need to increase high octane components is reduced if these technologies are used, the net benefit of low sulfur gasoline is even greater, because there are even fewer toxic compounds for the catalyst to have to convert.

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### Appendix D References

1. *EPA Staff Paper on Gasoline Sulfur Issues* (EPA420-R-98-004), May 1998.